Subject	Monday	Tuesday	Wednesday	Thursday	Friday
ACCRS:	se the periodic table as a model to predict the relative properties and trends (e.g., reactivity of metals; types of bonds formed, including ionic, covalent, and polar covalent; numbers of bonds formed; reactions with oxygen) of main group elements based on the patterns of valence electrons in atoms.	se the periodic table as a model to predict the relative properties and trends (e.g., reactivity of metals; types of bonds formed, including ionic, covalent, and polar covalent; numbers of bonds formed; reactions with oxygen) of main group elements based on the patterns of valence electrons in atoms.	se the periodic table as a model to predict the relative properties and trends (e.g., reactivity of metals; types of bonds formed, including ionic, covalent, and polar covalent; numbers of bonds formed; reactions with oxygen) of main group elements based on the patterns of valence electrons in atoms.	se the periodic table as a model to predict the relative properties and trends (e.g., reactivity of metals; types of bonds formed, including ionic, covalent, and polar covalent; numbers of bonds formed; reactions with oxygen) of main group elements based on the patterns of valence electrons in atoms.	se the periodic table as a model to predict the relative properties and trends (e.g., reactivity of metals; types of bonds formed, including ionic, covalent, and polar covalent; numbers of bonds formed; reactions with oxygen) of main group elements based on the patterns of valence electrons in atoms.
Before	Kahoot quiz	Data Set Quiz 1	Kahoot quiz	Checkpoint quiz 1	
During	Electron configuration practice	Lewis dot diagrams/ the octet rule	Chemical bonding nearpod	Molecules nearpod	Chemical formulas and nomenclature nearpod
After		activity	activity		activity
Desired Outcome	To learn how electrons are distributed in their energy shells around an atom	To show how to use lewis structures to represent the valence electrons of an atom	To show how atoms are bonded together to form molecules	To learn what molecules are and how they form	To learn the types of formulas and how to name some chemicals
Formative/ Summative	quiz	quiz	quiz	quiz	